

## SODIUM BICARBONATE FOR METABOLIC ACIDOSIS IN HOSPITAL USM

### INTRODUCTION

Sodium bicarbonate ( $\text{NaHCO}_3$ ) is used in the treatment of a wide variety of metabolic acidosis

### MECHANISM OF ACTION<sup>(2)</sup>

It provides bicarbonate ion which neutralizes hydrogen ion concentration and raises blood and urinary pH.

Availability <sup>5</sup>	IV $\text{NaHCO}_3$ 8.4%	Mist $\text{NaHCO}_3$ 8.4%	$\text{NaHCO}_3$ powder
Content <sup>5</sup>	1ml = 84mg of $\text{NaHCO}_3$ = 1mmol $\text{Na}^+$ & 1mmol $\text{HCO}^{3-}$		1g = 11.9mmol $\text{Na}^+$ & 11.9mmol $\text{HCO}^{3-}$
Indication <sup>2</sup>	Acute severe metabolic acidosis	Metabolic acidosis in patients with chronic kidney disease	
Dosing	<p><u>If acid-base status available<sup>(2,4)</sup>:</u></p> <p>Dosing based on <math>\text{HCO}^{3-}</math> deficit = <math>0.5 \times \text{weight (kg)} \times [\text{desired } \text{HCO}^{3-} - \text{measured } \text{HCO}^{3-} (\text{mmol/L})]</math></p> <p><u>If acid-base status not available<sup>(2,3,4)</sup>:</u></p> <p>Severe cases (e.g. cardiac arrest): 1 mmol/kg via slow IV, followed by 0.5 mmol/kg given at 10-minute intervals depending on individual arterial blood gases</p> <p>Less urgent cases: 2 to 5 mmol/kg IV infusion over 4 to 8 hours; subsequent doses should be based on patient's acid-base status</p>	<p>15 - 20 ml/ day in divided doses (max: 70ml/day)<sup>(2)</sup></p>	<p>1200mg – 1950mg/day in divided doses (max: 5850mg/day)<sup>(2)</sup></p> <p>Example: 500 mg TDS</p> <p>* In fluid restricted patient, <math>\text{NaHCO}_3</math> powder is preferred compared to mist <math>\text{NaHCO}_3</math> 8.4%</p>

## DILUTION AND ADMINISTRATION OF IV NaHCO<sub>3</sub> 8.4%

### LESS URGENT FORMS OF METABOLIC ACIDOSIS

#### Choice 1<sup>(6,7)</sup>

Based on this formula: **M1 x V1 = M2 x V2**

M1: Existing concentration (8.4%) ; V1: Volume to syringe out from 8.4%

M2: Intended concentration isotonic (1.5%) ; V2: Final volume

**Example:** To produce isotonic NaHCO<sub>3</sub> 1.5% from 10 ml NaHCO<sub>3</sub> 8.4%, how many diluent that we need?

$$(8.4\%) \times 10 \text{ ml} = (1.5\%) \times V2 \text{ ml}$$

$$V2 = (8.4 \times 10) / 1.5$$

$$= 56 \text{ ml [final volume]}$$

Therefore diluent needed is 56 ml - 10 ml = 46 ml

**Conclusion:** Dilute 1 ampoule (10 ml) NaHCO<sub>3</sub> 8.4% in 46 ml of diluent (WFI/NS/D5) to produce 56 ml NaHCO<sub>3</sub> 1.5% w/v (isotonic)

#### Choice 2<sup>(7)</sup>

An IV infusion can be prepared by adding 75ml sodium bicarbonate to 425mL D5W (288 mOsm/kg)

**Administer over 4 to 8 hours<sup>(3,9)</sup> (maximum rate of administration: 1 mEq/kg/hour)<sup>(2)</sup>**

### SEVERE METABOLIC ACIDOSIS (PH <7.1)

#### Choice 1<sup>(8)</sup> [Practiced in ICU medical ward]

**Undiluted** 50 mmol IV NaHCO<sub>3</sub> 8.4% (50ml) given **over 1 hour** in severe DKA.

\*Plasma pH, electrolytes and pCO<sub>2</sub> must be closely monitored and over-correction avoided

#### Choice 2<sup>(6,7)</sup>

**For emergency cases, e.g: cardiac arrest**

**Undiluted** 8.4% IV NaHCO<sub>3</sub> given by **slow IV injection over 5 min** (refer dosing cardiac arrest)

**Concentrations >1.26% should be given via a central line except in emergencies<sup>(8)</sup>**

#### References:

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